

# Tailoring olefin distribution via tuning rare earth metals in bifunctional Cu-RE/beta-zeolite catalysts for ethanol upgrading

## Background

- Bioethanol to middle distillate technologies have offered a unique solution to produce renewable aviation fuel for decarbonizing the hard-to-electrify sectors.
- Direct and selective production of  $C_{3+}$  alkene from bioethanol remains a critical challenge and important to produce renewable transportation fuels such as aviation biofuels.

## Approach

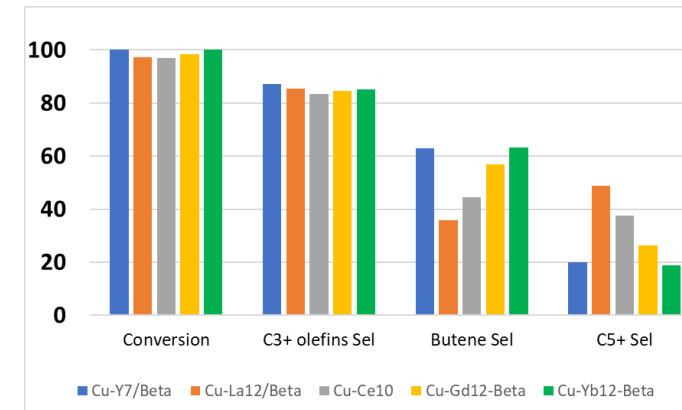
- The bimetallic Cu- and rare earth (RE)-doped DeAl-Beta zeolites (Cu-RE/Beta) catalysts were synthesized via solid state grinding
- Home-made microflow reactor was used to perform ethanol upgrading reaction and kinetic reaction rate measurement experiments
- Products were separated and analyzed using a gas chromatography

## Results

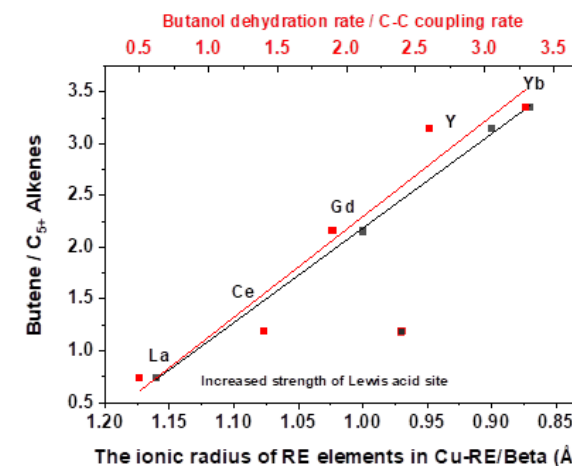
- Developed a series of Cu-RE/Beta catalysts that yield high  $C_{3+}$  alkene selectivity from ethanol upgrading (>80% selectivity at ~100% conversion, 623K)
- The formation rates of butene isomers to  $C_{5+}$  alkenes are linearly correlated with the strength of Lewis acidic RE identity, which follows the sequence of  $Yb_{12}/Beta > Y_7/Beta > Gd_{12}/Beta > Ce_{10}/Beta > La_{12}/Beta$
- RE selection plays the vital role in altering the rate of the key competitive reactions within the ethanol-to-alkenes reaction network, namely  $C_4$  alcohol dehydration and C-C chain growth, which dictate alkene product distributions

## Significance

- This work provides a feasible and promising method for tailoring alkene product distributions from ethanol upgrading, which is of notable significance to the generation of renewable fuels.



Ethanol conversions and olefin product selectivities over bimetallic Cu-RE/Beta



The relative ratio of butene to  $C_{5+}$  olefin as a function of the ionic radius of RE metals, simultaneously in line with the fraction of the initial rate of butanol dehydration to C-C coupling