

# Pentafluoropyridine with $^{19}\text{F}$ NMR offers a safe and cost-effective alternative to quantify phenolic groups in lignin and lignin-derived products

## Background

- Phenolic groups are important for both reactivity and material properties in lignin and lignin-derived products. A widely implemented method for quantifying phenolic groups in lignin is phosphitylation followed by  $^{31}\text{P}$  NMR spectroscopy. This straightforward method suffers from important drawbacks such as an expensive hazardous derivatization agent and degradation of the reagent and derivatized samples.

## Approach

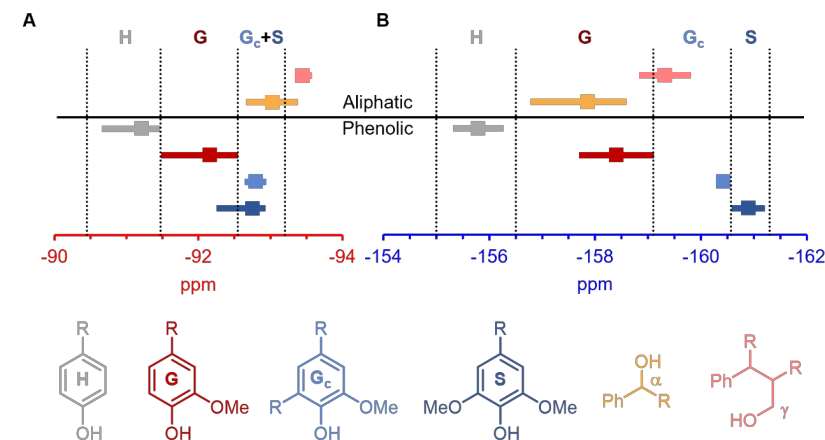
- We hypothesized that derivatizing lignins with pentafluoropyridine (PFP) followed by  $^{19}\text{F}$  NMR could offer a safer and cheaper method to quantify phenolics (\$6/g for PFP vs. \$217/g for the phosphitylation reagent). We screened approximately 30 lignin model compounds for their reactivity with PFP. From these results, we compared  $^{19}\text{F}$  NMR method to the  $^{31}\text{P}$  method for three lignins: poplar RCF oil, softwood kraft lignin, and corn stover milled wood lignin.

## Results

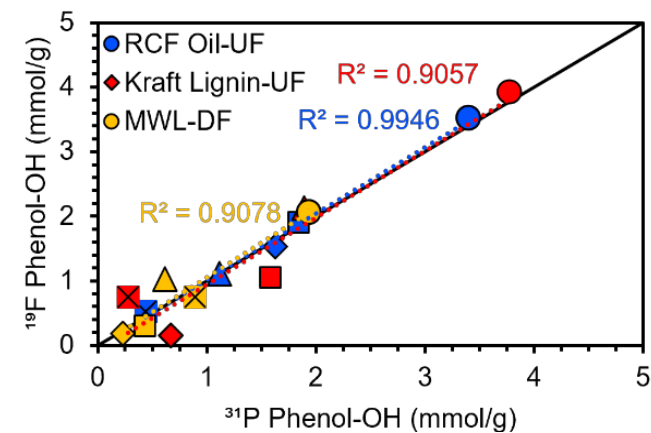
- Using a 40%  $\text{H}_2\text{O}/\text{DMSO}$  solvent allowed for complete and selective reaction with most model compounds studied. Although aliphatic hydroxyl groups may react with PFP, high selectivity to the phenolic reaction is obtained by using a short (5 min) reaction time. The chemical shift of the products allowed for differentiation of 5-substituted, S, G, and H moieties. Good agreement with  $^{31}\text{P}$  results for total phenol quantification was obtained for all three lignins studied. There were some differences for specific moieties: for softwood, the  $^{19}\text{F}$  method measured a lower amount of S units while for RCF oil, the S, G, and 5-substituted moieties were indistinguishable.

## Significance

- A comparable analytical method avoids the use of a dangerous reagent and improves safety – especially for less experienced chemists. The method supplements the tools available to characterize lignin hydroxyl groups. Furthermore, the higher derivatized sample stability in DMSO allows for future higher-throughput applications.



$^{19}\text{F}$  NMR chemical shift of phenolic moieties in 40%  $\text{H}_2\text{O}/\text{DMSO}$ .



Parity plot of  $^{31}\text{P}$  results versus  $^{19}\text{F}$  for poplar RCF oil, softwood kraft lignin, and corn stover milled wood lignin.

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